



IN THE UNITED STATES PATENT & TRADEMARK OFFICE

In Re the Application of:) Group Art Unit: Not yet assigned
Wang, et al.	Examiner: Not yet assigned
Serial No.: 10/581,418)
Filed: June 2, 2006) CERTIFICATE OF MAILING) I HEREBY CERTIFY THAT THIS CORRESPONDENCE IS
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Title: Method for Selective Reduction of Aromatic Compounds))) MICHELLE DE CORTE

INFORMATION DISCLOSURE STATEMENT

Mail Stop: Amendment Commissioner for Patents PO Box 1450 Alexandria, VA 22313-1450

Dear Sir:

In accordance with duties imposed by 37 CFR §1.56 and means for complying therewith according to 37 CFR §§1.97-1.98, the references listed on the attached Form PTO/SB/08A are called to the attention of the US Patent & Trademark Office in relation to the present application.

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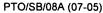
Since this correspondence is being submitted prior to a first Office Action on the merits, it is believed that no fee is due. However, if a fee is required for entry of this correspondence, please charge Deposit Account No. 13-1160.

Respectfully submitted,

Christine Rebman Cooke (Reg. No. 50,546) mailto:christine.cooke@tycohealthcare.com

Chrome P. Cooke

314-654-3939



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Complete if Known **Application Number** 10/581,418 INFORMATION DISCLOSURE June 2, 2006 Filing Date STATEMENT BY APPLICANT First Named Inventor WANG Art Unit Not yet assigned Not yet assigned **Examiner Name** (use as many sheets as necessary) 1618 WO/US 1 Attorney Docket Number of Sheet | 1

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		NON PATENT LITERATURE DOCUMENTS	T ²
Examiner Initials*	Cite No. ¹	Include name of the author (in CAPITAL LETTERS), title of the article (when appropriate), title of The item (book, magazine, journal, serial, symposium, catalog, etc.) date, page(s), volume-issue number(s), publisher, city and/or country where published.	
	A1	BIRCH, Reduction by Dissolving Metals. Part I., Chem. Soc. 1944, pp. 430-436	
	A2	BIRCH, The reduction of organic compounds by metal-ammonia solution, Quart. Rev. Chem. Soc. 1950, 4, pp. 69-93	
	A3	RADIDEAU, The metal-ammonia reduction of aromatic compounds, Tetrahedron 1989, 45, pp. 1579-1603	
	A4	MANDER, Partial reduction of aromatic rings by dissolving metals and by other methods, Comprehensive Organis synthesis 1991, 8, pp. 489-521	
	A5	BIRCH, The Birch reduction in organic synthesis, Pure & Appl. Chem. 1996, 68, pp. 553-555	
	A6	ALONSO et al., The NiCl ₂ .2H ₂ O-Li-arene combination as reducing system. 4. Dehalogenation of organic halides using the NiCl ₂ .2H ₂ O-Li-DTBB (cat.) combination, Tetrahedron 1999, 55, pp. 4441-4444	
	A7	BERKOWITZ, An efficient dechlorination method for 1,2,3,4-tetrachloro-5,5-dimethoxycyclopentadiene Diels-Alder adduct: Inverse addition-etheral Birch reduction condition, Synthesis 1990, 8, pp. 649-651	
	A8	BRYCE-SMITH et al., Reduction of organic halides. Chlorobenzene to benzene, Org. Synth	
	A9	ROSSI et al., On the dehydroxylation of phenols by cleavage of their diethyl phosphate esters with alkali methals in liquid ammonia, J. Org. Chem. 1973, 38, p. 2314	
	A10	WELCH et al., Reduction of aryl diethyl phosphates with titanium metal: a method for depaymenation of phenols, J. Org. Chem. 1978, 43, p. 4797	
	A11	BIRCH et al., Reaction mechanisms in reductions by metal-ammonia solutions, Tetrahedron	
	A12	KAISER, A Comparison of Methods Using Lithium/Amine and Birch Reduction Systems,	
	A13	BIRCH et al., Reductions by metal-ammonia solutions and related reagent, Advanced Organic Chemistry 1972, 8, pp. 1-65 XP009049761	
	A14	International Search Report dated June 28, 2005	

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